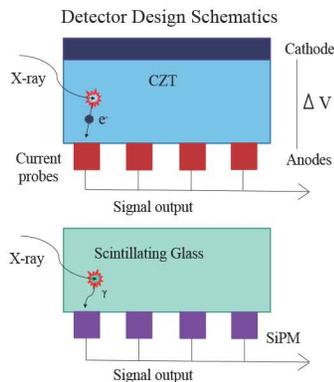


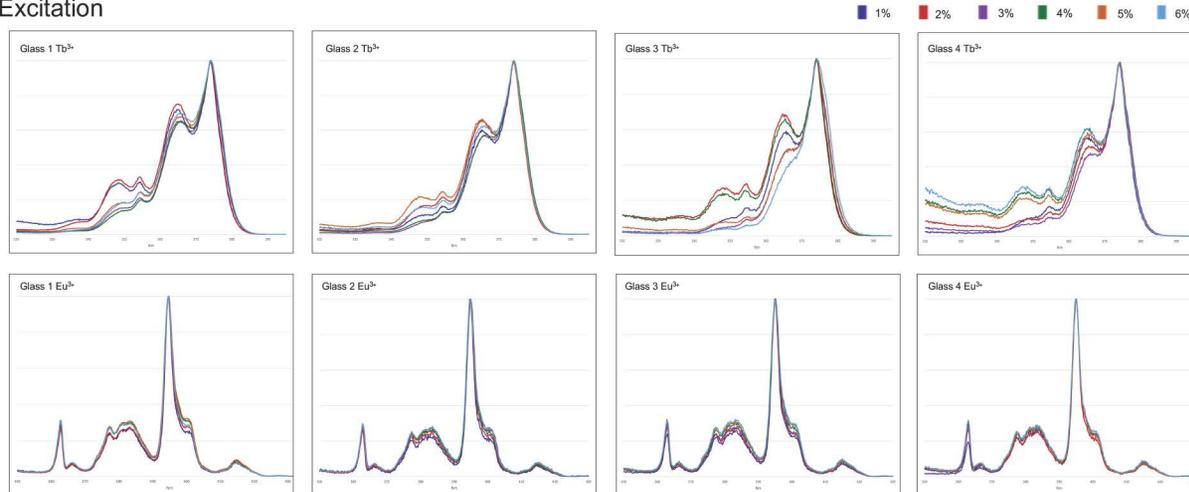
Introduction

X-ray detectors are commonly used for imaging purposes in medical physics. This study focuses on producing scintillating glasses that can be used as an efficient and cost effective alternative to the current market material for X-ray detectors- which is cadmium zinc telluride (CZT), a semiconducting crystal. Scintillators are materials which will absorb energy and consequently emit photons in the visible range. For use in a medical detector the scintillating material must have a number of special properties:

- The density must be high enough to be a good absorber for x-rays
- The glass must have a high degree of optical transparency for emission wavelength
- The emission wavelength should match perfectly with the high quantum efficiency region of the photodetectors
- The energy output must rival CZT
- It must be able to perform with high precision at relatively standard pressures and temperatures over time



Excitation



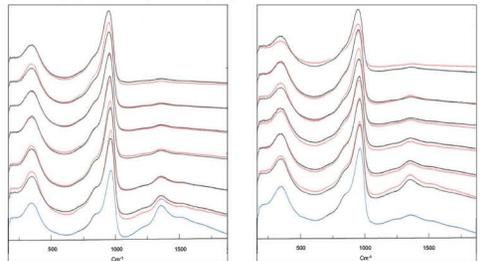
Glass Making Conditions

Glass samples composed of $X \text{ Gd}_2\text{O}_3 + Y \text{ WO}_3 + (1-X-Y) 2\text{H}_3\text{BO}_3$ were formed. If $X=.25$ then $Y=.25, .35, .45$ and if $X=.20$ then $Y=.20$. For each glass composition, an additional 12 glass samples were created containing between 1-6% Eu³⁺ or Tb³⁺. The amount of each reagent was varied to manipulate density and clarity while the amount of dopant was varied to maximize scintillation while avoiding self-quenching. Reagents were measured into an alumina crucible in the desired ratios and were mixed by hand for at least five minutes. The crucible was then placed into a furnace that was preheated to 1200°C and was left for 30 minutes to allow for sufficient melting. The molten glass was poured onto a room temperature iron plate and formed readily.

Raman Spectra

Glass 1
0.20 Gd₂O₃ - .20 WO₃ - .60 2H₃BO₃

Glass 2
0.25 Gd₂O₃ - .25 WO₃ - .50 2H₃BO₃



— Terbium — Europium

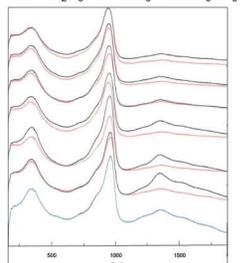
The peak occurring under 500 cm⁻¹ and the peak occurring just below 1000 cm⁻¹ shift slightly to the left.

• There is a clear decrease in the intensity of the peaks occurring near 1350 cm⁻¹; doping with Tb³⁺ shows a more gradual shift in this behavior than with Eu³⁺.

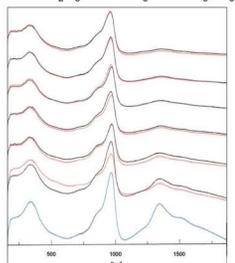
- The peak occurring under 500 cm⁻¹ and the peak occurring just below 1000 cm⁻¹ shift slightly to the left.
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This behavior indicates that the addition of Eu³⁺ and Tb³⁺ in the glass causes the preferential formation of tetrahedra (941) and Ln-O-Ln clusters (~300; where Ln=Gd, Eu, or Tb) in the glasses.

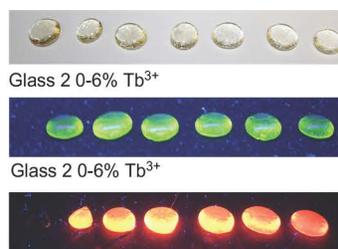
Glass 3
0.20 Gd₂O₃ - .20 WO₃ - .60 2H₃BO₃



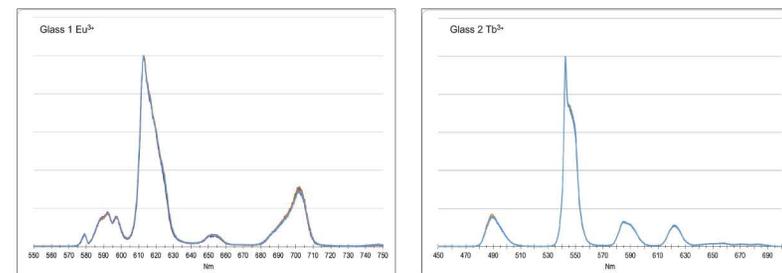
Glass 4
0.25 Gd₂O₃ - .45 WO₃ - .30 2H₃BO₃



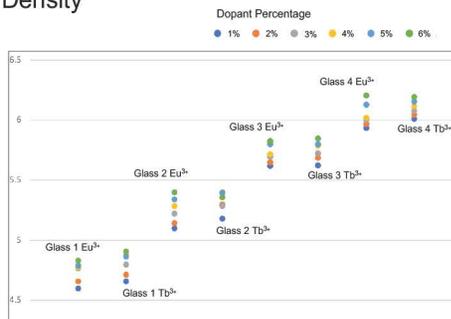
Emission



The optimal concentration of scintillating ions in the glass to produce the brightest signal appears to be 3%.



Density



Density studies show that the densities of the glasses increased with the addition of rare-earths. Previous studies struggled to reach densities of 6.0 g/cm³

Conclusion

Europium or Terbium doped gadolinium borotungstate glasses show very strong potential for use as a cheaper and more efficient alternative to CZT in x-ray detectors. The presence of scintillating ions within gadolinium borotungstate glasses have shown an obvious effect on the structure formed within the glasses. Additionally, as the dopant concentration in the glass increases, the density increases and a change is apparent in the raman spectra. Excitation data further suggests that the local environments surrounding the scintillators have an effect on the ratio of intensity of each emission band, and that there may exist a correlation between surrounding structure and excitation properties. Future studies will be conducted to determine the energy output of the glasses as well as the glass transition temperatures and further the stability of the glasses.

References and Acknowledgements

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